



STIC Search Report

EIC 1700

STIC Database Tracking Number: 196372

TO: Laura Weiner
Location: REM 6C83
Art Unit : 1745
July 26, 2006

Case Serial Number: 10/647541

From: Kathleen Fuller
Location: EIC 1700
REMSEN 4B28
Phone: 571/272-2505
Kathleen.Fuller@uspto.gov

Search Notes

BROAD SEARCH COVERING FORMULAS 1-8 AND ELECTROLYTE OR FROMULAS 1-8 AND FORMULA 9 AND ELECTROLYTE. ONLY 9 CA REFERENCES.

=> file reg
FILE 'REGISTRY' ENTERED AT 15:19:14 ON 26 JUL 2006
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STRUCTURE FILE UPDATES: 25 JUL 2006 HIGHEST RN 895581-37-0
DICTIONARY FILE UPDATES: 25 JUL 2006 HIGHEST RN 895581-37-0

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TSCA INFORMATION NOW CURRENT THROUGH January 6, 2006

Please note that search-term pricing does apply when conducting SmartSELECT searches.

REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

<http://www.cas.org/ONLINE/UG/regprops.html>

=> file hcapl
FILE 'HCAPLUS' ENTERED AT 15:19:18 ON 26 JUL 2006
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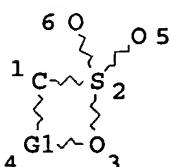
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FILE COVERS 1907 - 26 Jul 2006 VOL 145 ISS 5
FILE LAST UPDATED: 25 Jul 2006 (20060725/ED)

New CAS Information Use Policies, enter HELP USAGETERMS for details.

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d que
L3 STR



*query covers all claims
1165 structures*

REP G1=(1-10) A

NODE ATTRIBUTES:

CONNECT IS E1 RC AT 5
 CONNECT IS E1 RC AT 6
 DEFAULT MLEVEL IS ATOM
 DEFAULT ECLEVEL IS LIMITED

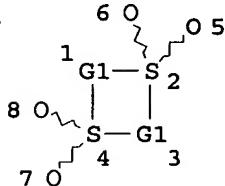
GRAPH ATTRIBUTES:

RSPEC I

NUMBER OF NODES IS 6

STEREO ATTRIBUTES: NONE

L5 1165 SEA FILE=REGISTRY SSS FUL L3
 L6 947 SEA FILE=REGISTRY ABB=ON L5 AND 1/S
 L7 883 SEA FILE=REGISTRY ABB=ON L6 NOT M/ELS
 L9 218 SEA FILE=REGISTRY ABB=ON L5 NOT L6
 L12 STR



883 structures per claim 11 with one S=0
 121 structures covering 2 S₁₁⁼⁰ in ring

REP G1=(1-5) A

NODE ATTRIBUTES:

CONNECT IS E1 RC AT 5
 CONNECT IS E1 RC AT 6
 CONNECT IS E1 RC AT 7
 CONNECT IS E1 RC AT 8
 DEFAULT MLEVEL IS ATOM
 DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RSPEC I

NUMBER OF NODES IS 8

STEREO ATTRIBUTES: NONE

L17 121 SEA FILE=REGISTRY SUB=L5 SSS FUL L12
 L18 114 SEA FILE=REGISTRY ABB=ON L17 NOT M/ELS
 L19 205 SEA FILE=HCAPLUS ABB=ON L9
 L20 7 SEA FILE=HCAPLUS ABB=ON L19(L) ELECTROLYTE?
 L21 134 SEA FILE=HCAPLUS ABB=ON L18
 L22 7 SEA FILE=HCAPLUS ABB=ON L21(L) ELECTROLYTE?
 L23 2653 SEA FILE=HCAPLUS ABB=ON L7
 L24 59 SEA FILE=HCAPLUS ABB=ON L19 AND L23
 L25 5 SEA FILE=HCAPLUS ABB=ON L24 AND ELECTROLYTE?
 L26 8 SEA FILE=HCAPLUS ABB=ON L20 OR L22 OR L25
 L27 9 SEA FILE=HCAPLUS ABB=ON (L19 OR L21) AND ELECTROLYTE?
 L28 9 SEA FILE=HCAPLUS ABB=ON L27 OR L26

=> d 128 1-9 bib abs ind hitstr

L28 ANSWER 1 OF 9 HCAPLUS COPYRIGHT 2006 ACS on STN
 AN 2005:888332 HCAPLUS
 DN 143:250968

Only 9 CA references for claim formulas
 1-8 and electrolyte
 or 1-8 and 9-~~10~~ and electrolyte

TI Electrolyte solutions containing cyclic sulfonate esters and secondary batteries using them

IN Kusachi, Yuki; Utsuki, Koji; Hasegawa, Etsuo

PA NEC Corp., Japan

SO Jpn. Kokai Tokkyo Koho, 23 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2005228631	A2	20050825	JP 2004-37003	20040213
PRAI	JP 2004-37003		20040213		

OS MARPAT 143:250968

AB The electrolyte solns. contain aprotic solvents and cyclic sulfonate esters with up to 10 sulfonate groups linked by C1-5 alkylene or fluoroalkylene groups. The solns. stabilize solid electrolyte interphase (SEI) films, resulting in batteries, e.g., secondary Li batteries, showing a long charge-discharge cycle life.

IC ICM H01M010-40

ICS H01M004-02; H01M004-38; H01M004-58

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

ST cyclic sulfonate ester electrolytic soln lithium battery

IT Carboxylic acids, uses

RL: DEV (Device component use); USES (Uses)

(aliphatic, esters; electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT Sulfonic acids, uses

RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)

(alkanesulfonic, anhydrides; electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT Sulfonic acids, uses

RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)

(cyclic sulfonic acid esters; electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT Ethers, uses

RL: DEV (Device component use); USES (Uses)

(cyclic; electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT Battery anodes

Battery cathodes

Battery electrolytes

Electrolytic solutions

(electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT Carbonates, uses

Ethers, uses

RL: DEV (Device component use); USES (Uses)

(electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT Secondary batteries

(lithium; electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT Lactones

RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)

(sultones, γ -; electrolytic solns. containing aprotic solvents and

cyclic sulfonate esters for secondary batteries)

IT Lactones
 RL: DEV (Device component use); USES (Uses)
 (γ -; electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT 7440-44-0, Carbon, uses
 RL: DEV (Device component use); USES (Uses)
 (amorphous, anode active mass; electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT 7439-93-2, Lithium, uses 7782-42-5, Graphite, uses
 RL: DEV (Device component use); USES (Uses)
 (anode active mass; electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

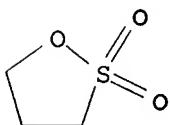
IT 12057-17-9, Lithium manganese oxide (LiMn₂O₄) 508200-28-0, Lithium manganese nickel titanium oxide (LiMn_{1.35}Ni_{0.5}Ti_{0.15}O₄)
 RL: DEV (Device component use); USES (Uses)
 (cathode active mass; electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 108-32-7,
 Propylene carbonate 872-36-6, Vinylene carbonate 3145-91-3, Lithium tetrachloroaurate 7791-03-9, Lithium perchlorate 14283-07-9, Lithium tetrafluoroborate 18424-17-4, Lithium hexafluoroantimonate 21324-40-3, Lithium hexafluorophosphate 29935-35-1, Lithium hexafluoroarsenate
 RL: DEV (Device component use); USES (Uses)
 (electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

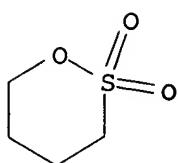
IT 126-33-0, Sulfolane 1120-71-4, 1,3-Propanesultone
 1633-83-6, 1,4-Butanesultone 28452-93-9D, Sulfolene, derivs.
 863198-22-5 863198-23-6 863198-24-7
 RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)
 (electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

IT 1120-71-4, 1,3-Propanesultone 1633-83-6,
 1,4-Butanesultone 863198-22-5 863198-23-6
 863198-24-7
 RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)
 (electrolytic solns. containing aprotic solvents and cyclic sulfonate esters for secondary batteries)

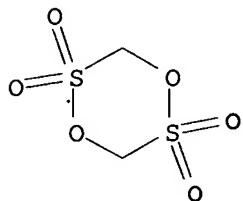
RN 1120-71-4 HCAPLUS
 CN 1,2-Oxathiolane, 2,2-dioxide (8CI, 9CI) (CA INDEX NAME)



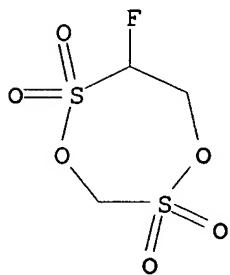
RN 1633-83-6 HCAPLUS
 CN 1,2-Oxathiane, 2,2-dioxide (8CI, 9CI) (CA INDEX NAME)



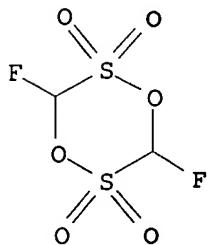
RN 863198-22-5 HCAPLUS
 CN 1,4,2,5-Dioxadithiane, 2,2,5,5-tetraoxide (9CI) (CA INDEX NAME)



RN 863198-23-6 HCAPLUS
 CN 1,4,2,5-Dioxadithiepane, 6-fluoro-, 2,2,5,5-tetraoxide (9CI) (CA INDEX NAME)



RN 863198-24-7 HCAPLUS
 CN 1,4,2,5-Dioxadithiane, 3,6-difluoro-, 2,2,5,5-tetraoxide (9CI) (CA INDEX NAME)



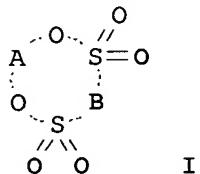
L28 ANSWER 2 OF 9 HCAPLUS COPYRIGHT 2006 ACS on STN
 AN 2005:822793 HCAPLUS
 DN 143:232664
 TI Electrolytes for secondary lithium batteries, and same batteries

IN Utsuki, Koji; Kusachi, Yuki
 PA NEC Corp., Japan
 SO Jpn. Kokai Tokkyo Koho, 24 pp.
 CODEN: JKXXAF

DT Patent
 LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2005222846	A2	20050818	JP 2004-30661	20040206
PRAI	JP 2004-30661		20040206		
OS	MARPAT 143:232664				
GI					



AB The electrolytes contain nonprototic solvents and 0.1-5.0 weight% of cyclic disulfonic acid esters, and the content of chlorine in the electrolytes is suppressed to <150 ppm. Preferably, the esters are expressed by I [A = (branched) (substituted) C1-5 alkylene, carbonyl, sulfinyl, (branched) perfluoroalkylene, etc.; B = (branched) (substituted) alkylene]. The esters provide protective films on electrode so as to prevent the electrodes from reaction with electrolyte solvents, so that secondary lithium batteries employing the electrolytes show high storage stability, and excellent charge-discharge cycling performance.

IC ICM H01M010-40

ICS H01M004-02; H01M004-38; H01M004-48; H01M004-58

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
Section cross-reference(s): 28ST lithium battery electrolyte cyclic disulfonic acid ester;
alkylenedisulfonic acid cyclic ester battery electrolyte

IT Carbonates, uses

Ethers, uses

RL: DEV (Device component use); USES (Uses)
(electrolyte components; lithium secondary battery
electrolytes containing cyclic disulfonic acid esters)

IT Carboxylic acids, uses

RL: DEV (Device component use); USES (Uses)
(esters, aliphatic, electrolyte components; lithium secondary
battery electrolytes containing cyclic disulfonic acid esters)

IT Battery electrolytes

(lithium secondary battery electrolytes containing cyclic
disulfonic acid esters)

IT Secondary batteries

(lithium; lithium secondary battery electrolytes containing
cyclic disulfonic acid esters)

IT Lactones

RL: DEV (Device component use); USES (Uses)
(γ -, aliphatic, electrolyte components; lithium secondary
battery electrolytes containing cyclic disulfonic acid esters)

IT 7791-03-9, Lithium perchlorate 14024-11-4, Aluminum lithium chloride

(AlLiCl₄) 14283-07-9, Lithium tetrafluoroborate 18424-17-4, Lithium hexafluoroantimonate 21324-40-3, Lithium hexafluorophosphate 29935-35-1, Lithium hexafluoroarsenate
 RL: DEV (Device component use); USES (Uses)
 (electrolyte components; lithium secondary battery
 electrolytes containing cyclic disulfonic acid esters)

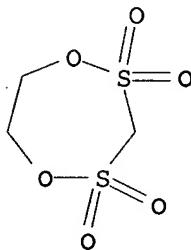
IT 99591-73-8P 99591-74-9P
 RL: DEV (Device component use); IMF (Industrial manufacture); PREP
 (Preparation); USES (Uses)
 (electrolyte components; lithium secondary battery
 electrolytes containing cyclic disulfonic acid esters)

IT 5799-68-8P, Methanedisulfonyl dichloride 71608-87-2P
 RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
 (Reactant or reagent)
 (in preparation of cyclic disulfonic acid esters for lithium secondary
 battery electrolytes)

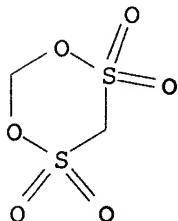
IT 75-11-6, Diiodomethane 107-21-1, Ethylene glycol, reactions 534-16-7,
 Silver carbonate 7790-94-5, Chlorosulfonic acid 10025-87-3, Phosphorus
 oxychloride
 RL: RCT (Reactant); RACT (Reactant or reagent)
 (in preparation of cyclic disulfonic acid esters for lithium secondary
 battery electrolytes)

IT 99591-73-8P 99591-74-9P
 RL: DEV (Device component use); IMF (Industrial manufacture); PREP
 (Preparation); USES (Uses)
 (electrolyte components; lithium secondary battery
 electrolytes containing cyclic disulfonic acid esters)

RN 99591-73-8 HCPLUS
 CN 1,5,2,4-Dioxadithiepane, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



RN 99591-74-9 HCPLUS
 CN 1,5,2,4-Dioxadithiane, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



L28 ANSWER 3 OF 9 HCPLUS COPYRIGHT 2006 ACS on STN
 AN 2005:547843 HCPLUS
 DN 143:81122

TI lithium secondary battery

IN Miyachi, Mariko; Utsugi, Koji; Kusachi, Yuki; Yamamoto, Hironori

PA NEC Corporation, Japan

SO PCT Int. Appl., 95 pp.

CODEN: PIXXD2

DT Patent

LA Japanese

FAN.CNT 3

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005057715	A1	20050623	WO 2004-JP18715	20041215
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
	JP 2005203341	A2	20050728	JP 2004-317279	20041029
	JP 2005203342	A2	20050728	JP 2004-317281	20041029
	JP 2005203343	A2	20050728	JP 2004-317297	20041029
	JP 2006156314	A2	20060615	JP 2004-363498	20041215
	JP 2006156315	A2	20060615	JP 2004-363502	20041215
PRAI	JP 2003-416516	A	20031215		
	JP 2004-317280	A	20041029		
	JP 2004-317298	A	20041029		
	JP 2004-317278	A	20041029		
	JP 2004-317299	A	20041029		

OS MARPAT 143:81122

AB The present invention aims to provide a lithium secondary battery with excellent characteristics such as energy d. and electromotive force, which is also

excellent in cycle life and shelf life stability. Disclosed is a secondary battery comprising at least a pos. electrode, a neg. electrode and an electrolyte solution wherein the neg. electrode contains a metal, metalloid or oxide, which adsorbs/desorbs an alkali metal or alkaline earth metal, and a carbon material as the neg. electrode active material, and the electrolyte solution contains a non-protonic solvent wherein at least an electrolyte is dissolved and a chain disulfone compound

IC ICM H01M010-40

ICS H01M004-02; H01M004-38; H01M004-48; H01M004-58

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

ST lithium secondary battery anode active substance electrolyte additive disulfone

IT Battery anodes

(anode active substances for)

IT Secondary batteries

(lithium; additives for)

IT Battery electrolytes

(nonaq.; disulfone additives for)

IT 872-36-6, Vinylene carbonate 1120-71-4, Propane sultone

2997-54-8 6330-39-8 22063-27-0 22063-28-1 23601-06-1

99591-74-9 152949-20-7 500878-47-7 855472-38-7 855472-43-4

855472-46-7

RL: MOA (Modifier or additive use); USES (Uses)

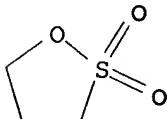
(additives for lithium non-aqueous electrolyte solution)

IT 1303-86-2, Boron oxide (B₂O₃), uses 1309-37-1, Ferric oxide, uses
 1314-56-3, Phosphorus oxide (P₂O₅), uses 7429-90-5, Aluminum, uses
 7439-89-6, Iron, uses 7439-92-1, Lead, uses 7440-02-0, Nickel, uses
 7440-21-3, Silicon, uses 7440-22-4, Silver, uses 7440-31-5, Tin, uses
 7440-32-6, Titanium, uses 7440-36-0, Antimony, uses 7440-50-8, Copper,
 uses 7440-56-4, Germanium, uses 7782-42-5, Graphite, uses
 12023-55-1, Iron silicide (Fe₃Si₇) 12031-95-7, Lithium titanium oxide
 (Li₄Ti₅O₁₂) 12036-84-9, Tungsten oxide (W₂O₅) 12042-55-6, Aluminum
 silicide (Al₂Si) 12334-14-4, Tin silicide (Sn₃Si) 18282-10-5, Tin
 dioxide 21651-19-4, Tin monoxide 39445-33-5 53095-76-4, Lithium
 silicide (Li₂Si) 113443-18-8, Silicon oxide (SiO) 160479-36-7, Lithium
 tin oxide 178958-56-0, Lithium silicon oxide 855472-17-2, Iron
 silicide (Fe₂Si₁₉) 855472-21-8, Aluminum nickel silicide (Al₉Ni₂Si₁₀)
 855472-26-3, Tin titanium silicide (SnTi₁₈Si) 855475-31-9
 RL: TEM (Technical or engineered material use); USES (Uses)
 (anode active substance for lithium secondary batteries)

IT 1120-71-4, Propane sultone 99591-74-9
 RL: MOA (Modifier or additive use); USES (Uses)
 (additives for lithium non-aqueous electrolyte solution)

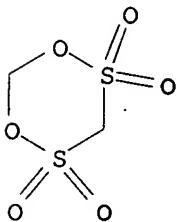
RN 1120-71-4 HCPLUS

CN 1,2-Oxathiolane, 2,2-dioxide (8CI, 9CI) (CA INDEX NAME)



RN 99591-74-9 HCPLUS

CN 1,5,2,4-Dioxadithiane, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



RE.CNT 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 4 OF 9 HCPLUS COPYRIGHT 2006 ACS on STN

AN 2005:547842 HCPLUS

DN 143:81121

TI Electrolyte solution for secondary lithium battery and the
 battery

IN Utsugi, Koji; Kusachi, Yuki; Katou, Tsuyoshi

PA NEC Corporation, Japan

SO PCT Int. Appl., 55 pp.

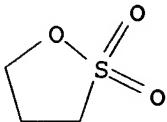
CODEN: PIXXD2

DT Patent

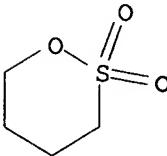
LA Japanese

FAN.CNT 3

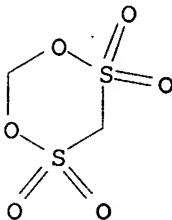
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005057714	A1	20050623	WO 2004-JP18698	20041215
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
	JP 2005203341	A2	20050728	JP 2004-317279	20041029
	JP 2005203342	A2	20050728	JP 2004-317281	20041029
	JP 2005203343	A2	20050728	JP 2004-317297	20041029
	CN 1757134	A	20060405	CN 2004-80005513	20041215
	JP 2006156314	A2	20060615	JP 2004-363498	20041215
	JP 2006156315	A2	20060615	JP 2004-363502	20041215
PRAI	JP 2003-416516	A	20031215		
	JP 2004-317301	A	20041029		
	JP 2004-317278	A	20041029		
	JP 2004-317299	A	20041029		
OS	MARPAT 143:81121				
AB	The electrolyte solution has an electrolyte dissolved in an aprotic solvent and contains a disulfone R3-SO2-CR1R4-SO2-R2, where R1-4 are various hydrocarbon groups which may also contain F, O, S, and N.				
IC	ICM H01M010-40 ICS H01M004-02; H01M004-38; H01M004-58; H01G009-038; H01L031-04; H01M014-00				
CC	52-2 (Electrochemical, Radiational, and Thermal Energy Technology)				
ST	secondary lithium battery electrolyte disulfone additive				
IT	Battery electrolytes (electrolyte solns. containing disulfone additives for secondary lithium batteries)				
IT	96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 108-32-7, Propylene carbonate 21324-40-3, Lithium hexafluorophosphate				
	RL: DEV (Device component use); USES (Uses) (electrolyte solns. containing disulfone additives for secondary lithium batteries)				
IT	872-36-6, Vinylene carbonate 1120-71-4, 1,3-Propanesultone 1633-83-6, 1,4-Butanesultone 2997-54-8 6330-39-8 22063-27-0 22063-28-1 23601-06-1 99591-74-9 152949-20-7 500878-47-7 855472-38-7 855472-43-4 855472-46-7				
	RL: MOA (Modifier or additive use); USES (Uses) (electrolyte solns. containing disulfone additives for secondary lithium batteries)				
IT	1120-71-4, 1,3-Propanesultone 1633-83-6, 1,4-Butanesultone 99591-74-9				
	RL: MOA (Modifier or additive use); USES (Uses) (electrolyte solns. containing disulfone additives for secondary lithium batteries)				
RN	1120-71-4 HCPLUS				
CN	1,2-Oxathiolane, 2,2-dioxide (8CI, 9CI) (CA INDEX NAME)				



RN 1633-83-6 HCPLUS
 CN 1,2-Oxathiane, 2,2-dioxide (8CI, 9CI) (CA INDEX NAME)



RN 99591-74-9 HCPLUS
 CN 1,5,2,4-Dioxadithiane, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



RE.CNT 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 5 OF 9 HCPLUS COPYRIGHT 2006 ACS on STN

AN 2005:283747 HCPLUS

DN 142:358030

TI Secondary nonaqueous electrolyte battery

IN Kusachi, Yuki; Utsugi, Koji

PA NEC Corporation, Japan

SO PCT Int. Appl., 34 pp.

CODEN: PIXXD2

DT Patent

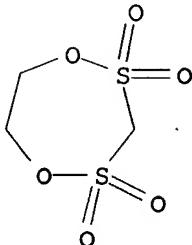
LA Japanese

FAN.CNT 1

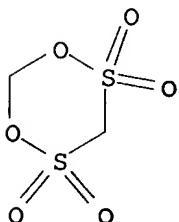
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005029613	A1	20050331	WO 2004-JP11534	20040811
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE,				

SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE,
SN, TD, TG

PRAI JP 2003-322968 A 20030916
 AB The battery has an anode whose surface is provided with a substance having a peak at 162.9-164.0 eV according to XPS anal.
 IC ICM H01M004-02
 ICS H01M010-40; H01M004-38; H01M004-58
 CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
 ST secondary battery anode surface electrolyte deposition XPS characteristic; battery electrolyte org sulfur contg compd
 IT Battery anodes
 Battery electrolytes
 (electrolytes having organic S containing compds. for secondary lithium batteries)
 IT Secondary batteries
 (lithium; electrolytes having organic S containing compds. for secondary lithium batteries)
 IT 7440-44-0, Carbon, uses
 RL: DEV (Device component use); USES (Uses)
 (amorphous; electrolytes having organic S containing compds. for secondary lithium batteries)
 IT 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 108-32-7,
 Propylene carbonate 12057-17-9, Lithium manganese oxide (LiMn₂O₄)
 21324-40-3, Lithium hexafluorophosphate 99591-73-8
 99591-74-9
 RL: DEV (Device component use); USES (Uses)
 (electrolytes having organic S containing compds. for secondary lithium batteries)
 IT 99591-73-8 99591-74-9
 RL: DEV (Device component use); USES (Uses)
 (electrolytes having organic S containing compds. for secondary lithium batteries)
 RN 99591-73-8 HCPLUS
 CN 1,5,2,4-Dioxadithiepane, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



RN 99591-74-9 HCPLUS
 CN 1,5,2,4-Dioxadithiane, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



RE.CNT 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

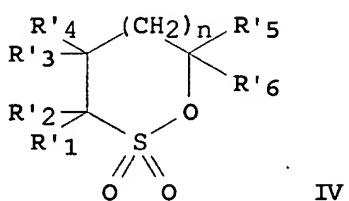
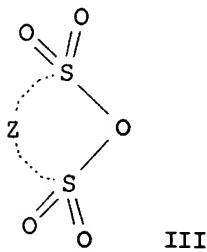
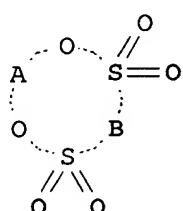
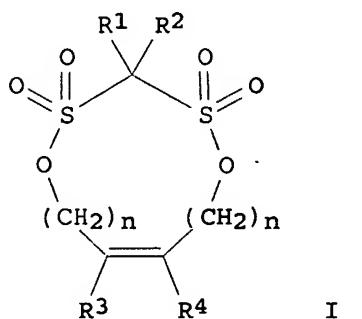
L28 ANSWER 6 OF 9 HCAPLUS COPYRIGHT 2006 ACS on STN
AN 2004:823614 HCAPLUS
DN 141:334876
TI Electrolyte solution for secondary battery and the battery
IN Kusachi, Yuki; Utsuki, Koji
PA NEC Corp., Japan
SO Jpn. Kokai Tokkyo Koho, 27 pp.
CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

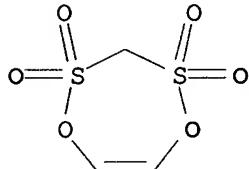
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2004281325	A2	20041007	JP 2003-74054	20030318
PRAI JP 2003-74054		20030318		
OS MARPAT 141:334876				
GI				



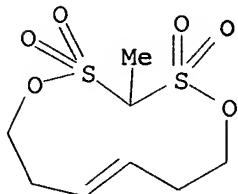
AB The electrolyte solution contains an aprotic solvent and an unsatd. cyclic disulfonate ester I, where R1-R4 = H, Me, Et, or halogen and n = integer 0-2. The electrolyte solution may also contain II [A = (substituted) C1-5 (fluoro)alkylene, carbonyl, sulfinyl, or bivalent C2-6 group containing ether bond connected (fluoro)alkylene units; B = (substituted) alkylene group], III [Z = (substituted) C2-4 alkylene, alkenylene, aromatic or heterocyclic group], or IV (n = integer 0-2, R'1-R'6 = H C1-12 alkyl, C3-6 cycloalkyl, or C6-12 aryl group). The battery is a secondary Li battery.

IC ICM H01M010-40
ICS H01M004-58

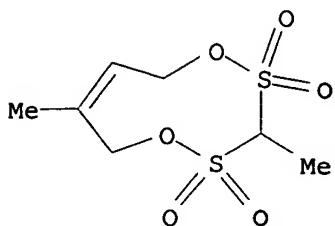
CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
 ST secondary lithium battery electrolyte soln cyclic disulfonate ester
 IT Battery electrolytes
 (electrolyte solns. containing cyclic disulfonate esters and other additives for secondary lithium batteries)
 IT 769973-24-2 769973-25-3 769973-26-4
 769973-27-5
 RL: MOA (Modifier or additive use); USES (Uses)
 (cyclic disulfonate ester containing secondary lithium battery electrolyte solns.)
 IT 872-36-6, Vinylene carbonate 1120-71-4, 1,3-Propanesultone
 14913-52-1, Neodymium ion (Nd³⁺), uses 18472-30-5, Erbium ion (Er³⁺), uses 22541-18-0, Europium ion (Eu³⁺), uses 22541-22-6, Holmium ion (Ho³⁺), uses 259194-36-0 259194-40-6 634598-36-0 634598-37-1
 RL: MOA (Modifier or additive use); USES (Uses)
 (electrolyte solns. containing cyclic disulfonate esters and other additives for secondary lithium batteries)
 IT 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 21324-40-3,
 Lithium hexafluorophosphate 132843-44-8
 RL: DEV (Device component use); USES (Uses)
 (electrolyte solns. containing cyclic disulfonate esters for secondary lithium batteries)
 IT 769973-24-2 769973-25-3 769973-26-4
 769973-27-5
 RL: MOA (Modifier or additive use); USES (Uses)
 (cyclic disulfonate ester containing secondary lithium battery electrolyte solns.)
 RN 769973-24-2 HCAPLUS
 CN 1,5,2,4-Dioxadithiepin, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



RN 769973-25-3 HCAPLUS
 CN 1,5-Dioxa-2,4-dithiacycloundec-8-ene, 3-methyl-, 2,2,4,4-tetraoxide (9CI)
 (CA INDEX NAME)

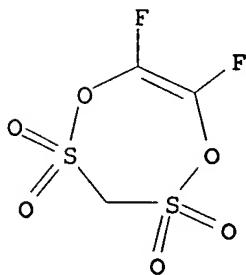


RN 769973-26-4 HCAPLUS
 CN 1,5,2,4-Dioxadithionin, 6,9-dihydro-3,7-dimethyl-, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



RN 769973-27-5 HCPLUS

CN 1,5,2,4-Dioxadithiepin, 6,7-difluoro-, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)

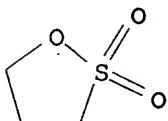


IT 1120-71-4, 1,3-Propanesultone

RL: MOA (Modifier or additive use); USES (Uses)
(electrolyte solns. containing cyclic disulfonate esters and other additives for secondary lithium batteries)

RN 1120-71-4 HCPLUS

CN 1,2-Oxathiolane, 2,2-dioxide (8CI, 9CI) (CA INDEX NAME)



L28 ANSWER 7 OF 9 HCPLUS COPYRIGHT 2006 ACS on STN

AN 2004:181920 HCPLUS

DN 140:184814

TI Electrolyte solution for secondary battery

IN Utsugi, Koji; Kusachi, Yuki; Yamazaki, Ikiko

PA NEC Corporation, Japan

SO Eur. Pat. Appl., 35 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP'1394888	A1	20040303	EP 2003-90268	20030822
	EP 1394888	B1	20060412		

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,

applicant

IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK

JP 2004281368	A2	20041007	JP 2003-289432	20030807
US 2004043300	A1	20040304	<u>US 2003-647541</u>	20030826
KR 2004019994	A	20040306	KR 2003-59849	20030828
CN 1495959	A	20040512	CN 2003-132755	20030829

PRAI JP 2002-250441 A 20020829
 JP 2003-52588 A 20030228
 JP 2003-289432 A 20030807

AB The present invention provides a technol. of inhibiting the decomposition of the solvent of the electrolyte solution for a secondary battery. Further, the present invention provides a technol. of prohibiting the resistance increase of a secondary battery and improving the storage properties such as improving the capacity retention ratio. An electrolyte solution comprising non-proton solvent and cyclic sulfonic ester including at least two sulfonyl groups may be used.

IC ICM H01M010-40
 ICS H01M006-16

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
 ST electrolyte soln secondary battery

IT Ethers, uses
 RL: DEV (Device component use); USES (Uses)
 (cyclic; electrolyte solution for secondary battery)

IT Battery electrolytes
 (electrolyte solution for secondary battery)

IT Ethers, uses
 Rare earth complexes
 Transition metal complexes
 RL: DEV (Device component use); USES (Uses)
 (electrolyte solution for secondary battery)

IT Carboxylic acids, uses
 RL: DEV (Device component use); USES (Uses)
 (esters, aliphatic; electrolyte solution for secondary battery)

IT Sulfonic acids, uses
 RL: DEV (Device component use); USES (Uses)
 (esters, cyclic; electrolyte solution for secondary battery)

IT Secondary batteries
 (lithium; electrolyte solution for secondary battery)

IT Lactones
 RL: DEV (Device component use); USES (Uses)
 (γ -; electrolyte solution for secondary battery)

IT 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 108-32-7, Propylene carbonate 463-79-6D, Carbonic acid, ester, cyclic 463-79-6D, Carbonic acid, ester, linear 497-62-1 872-36-6, Vinylene carbonate 7429-90-5, Aluminum, uses 7439-93-2, Lithium, uses 7440-00-8D, Neodymium, complex 7440-44-0, Carbon, uses 7440-52-0D, Erbium, complex 7440-53-1D, Europium, complex 7440-60-0D, Holmium, complex 7782-42-5, Graphite, uses 7791-03-9, Lithium perchlorate 12057-17-9, Lithium manganese oxide limn₂O₄ 14024-11-4, Lithium tetrachloroaluminate 14283-07-9, Lithium tetrafluoroborate 18424-17-4, Lithium hexafluoroantimonate 21324-40-3, Lithium hexafluorophosphate 29935-35-1, Lithium hexafluoroarsenate 113066-89-0, Cobalt lithium nickel oxide Co_{0.2}LiNi_{0.8}O₂ 132843-44-8
 RL: DEV (Device component use); USES (Uses)
 (electrolyte solution for secondary battery)

IT 1120-71-4, 1,3-Propanesultone 14913-52-1, Neodymium(3+), uses 18472-30-5, Erbium(3+), uses 22541-18-0, Europium(3+), uses 22541-22-6, Holmium(3+), uses 37181-39-8, Triflate 99591-73-8 99591-74-9 99591-80-7 259194-36-0 259194-40-6 634598-36-0 634598-37-1 659737-87-8 659737-88-9 659737-89-0 659737-90-3

RL: MOA (Modifier or additive use); USES (Uses)
 (electrolyte solution for secondary battery)

IT 1120-71-4, 1,3-Propanesultone 99591-73-8

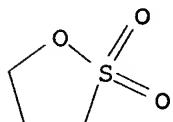
99591-74-9 99591-80-7 659737-87-8

659737-88-9 659737-89-0 659737-90-3

RL: MOA (Modifier or additive use); USES (Uses)
 (electrolyte solution for secondary battery)

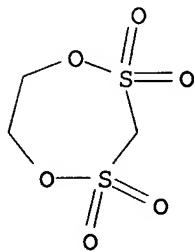
RN 1120-71-4 HCPLUS

CN 1,2-Oxathiolane, 2,2-dioxide (8CI, 9CI) (CA INDEX NAME)



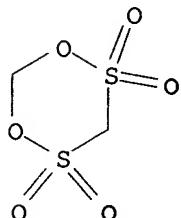
RN 99591-73-8 HCPLUS

CN 1,5,2,4-Dioxadithiepane, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



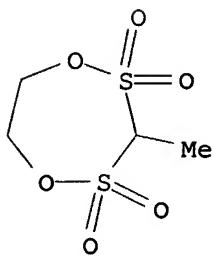
RN 99591-74-9 HCPLUS

CN 1,5,2,4-Dioxadithiane, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



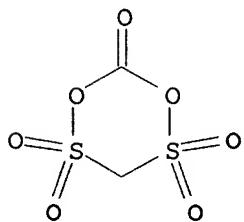
RN 99591-80-7 HCPLUS

CN 1,5,2,4-Dioxadithiepane, 3-methyl-, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



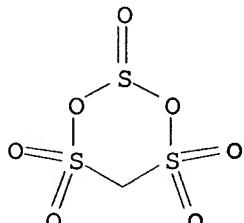
RN 659737-87-8 HCAPLUS

CN 1,5,2,4-Dioxadithian-6-one, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



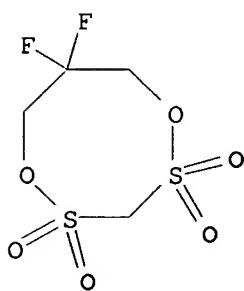
RN 659737-88-9 HCAPLUS

CN 1,3,2,4,6-Dioxatrithiiane, 2,4,4,6,6-pentaoxide (9CI) (CA INDEX NAME)



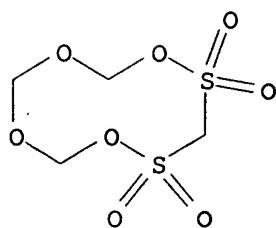
RN 659737-89-0 HCAPLUS

CN 1,5,2,4-Dioxadithiocane, 7,7-difluoro-, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



RN 659737-90-3 HCAPLUS

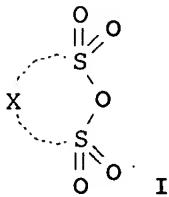
CN 1,5,7,9,2,4-Tetroxadithiepane, 2,2,4,4-tetraoxide (9CI) (CA INDEX NAME)



RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L28 ANSWER 8 OF 9 HCAPLUS COPYRIGHT 2006 ACS on STN
 AN 2004:57903 HCAPLUS
 DN 140:131080
 TI Electrolyte solution for the secondary battery and the battery
using the solution
 IN Utsuki, Koji; Mori, Mitsuhiro
 PA NEC Corp., Japan
 SO Jpn. Kokai Tokkyo Koho, 24 pp.
 CODEN: JKXXAF
 DT Patent
 LA Japanese
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2004022336	A2	20040122	JP 2002-175648	20020617
PRAI JP 2002-175648		20020617		
GI				



AB The electrolyte solution has a sulfonic acid anhydride I [X =
(substituted) C2-4 alkylene, (substituted) C2-4 alkenyl, or (substituted)
aromatic ring] in an aprotic solvent. The battery has a cathode, an anode,
and the above electrolyte solution
 IC ICM H01M010-40
 ICS H01M004-02; H01M004-58
 CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
 ST secondary battery electrolyte sulfonic acid anhydride
 IT Battery electrolytes
 Secondary batteries
 (electrolyte solns. containing sulfonic acid anhydrides for
 secondary batteries)
 IT 7440-44-0, Carbon, uses
 RL: DEV (Device component use); USES (Uses)
 (amorphous; anode; electrolyte solns. containing sulfonic acid
 anhydrides for secondary batteries)
 IT 7439-93-2, Lithium, uses 7782-42-5, Graphite, uses

RL: DEV (Device component use); USES (Uses)
 (anode; electrolyte solns. containing sulfonic acid anhydrides
 for secondary batteries)

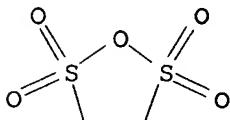
IT 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 108-32-7,
 Propylene carbonate 12057-17-9, Lithium manganese oxide (LiMn₂O₄)
 21324-40-3, Lithium hexafluorophosphate 33356-82-0 132843-44-8
 RL: DEV (Device component use); USES (Uses)
 (electrolyte solns. containing sulfonic acid anhydrides for
 secondary batteries)

IT 872-36-6, Vinylene carbonate 4378-87-4 76076-58-9
 82727-20-6 259194-36-0 259194-40-6 634598-36-0 634598-37-1
 648922-25-2 648922-26-3 648922-27-4
 RL: MOA (Modifier or additive use); USES (Uses)
 (electrolyte solns. containing sulfonic acid anhydrides for
 secondary batteries)

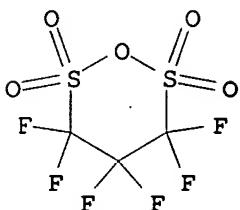
IT 4378-87-4 82727-20-6
 RL: MOA (Modifier or additive use); USES (Uses)
 (electrolyte solns. containing sulfonic acid anhydrides for
 secondary batteries)

RN 4378-87-4 HCPLUS

CN 1,2,5-Oxadithiolane, 2,2,5,5-tetraoxide (9CI) (CA INDEX NAME)



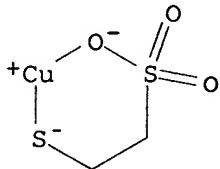
RN 82727-20-6 HCPLUS
 CN 1,2,6-Oxadithiane, 3,3,4,4,5,5-hexafluoro-, 2,2,6,6-tetraoxide (9CI) (CA
 INDEX NAME)



L28 ANSWER 9 OF 9 HCPLUS COPYRIGHT 2006 ACS on STN
 AN 1990:167909 HCPLUS
 DN 112:167909
 TI Bath for electrodeposition of smooth copper layers
 IN Loewe, Holger; Schmidt, Helge; Kiessling, Sabine; Vieweger, Ulrich;
 Schmidt, Cordt; Liebscher, Heinz; Kurz, Stefan
 PA Technische Hochschule Ilmenau, Ger. Dem. Rep.
 SO Ger. (East), 7 pp.
 CODEN: GEXXA8
 DT Patent
 LA German
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI DD 269068 A3 19890621 DD 1987-301490 19870406
 PRAI DD 1987-301490 19870406
 OS MARPAT 112:167909
 AB The bath contains a H₂SO₄-containing CuSO₄ electrolyte and a S-containing organic H₂O-soluble additive. The additive, at a concentration of 0.01-100 mg/L, permits a constant deposition of smooth or bright and ductile Cu layers when used with inert anode materials and at c.d. ≤1000 A/dm².
 IC ICM C25D003-38
 CC 72-8 (Electrochemistry)
 ST smooth copper layer electrodeposition bath
 IT 126285-75-4 126397-51-1 126397-53-3 126397-54-4
 126397-55-5
 RL: PRP (Properties)
 (electrodeposition of smooth copper layers from baths containing)
 IT 7440-50-8, Copper, uses and miscellaneous
 RL: USES (Uses)
 (electrodeposition of smooth layers of, bath for)
 IT 126285-75-4
 RL: PRP (Properties)
 (electrodeposition of smooth copper layers from baths containing)
 RN 126285-75-4 HCPLUS
 CN Cuprate(1-), [2-mercaptoethanesulfonato(2-)O₁S₂]⁻, potassium (9CI) (CA INDEX NAME)



● K⁺

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